

# College Physics Fourth Edition Solutions Manual

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E1-22 First nd the "logarithmic average" by  $\log_d av = \frac{1}{2} \log(2 \cdot 1026) + \log(1 \cdot 10 \cdot 15) = \frac{1}{2} \log 2 \cdot 1026 \cdot 1 \cdot 10 \cdot 15 = \log p \cdot 2 \cdot 1011$  Solve, and  $d av = 450 \text{ km}$  E1-23 The number of atoms is given by  $(1 \text{ kg}) / (1.00783 \cdot 1.661 \cdot 10^{-27} \text{ kg})$ , or  $5.974 \cdot 10^{26}$  atoms

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